

10050

DRAFT

Ilmenite Basalt (low K)

114.5 grams

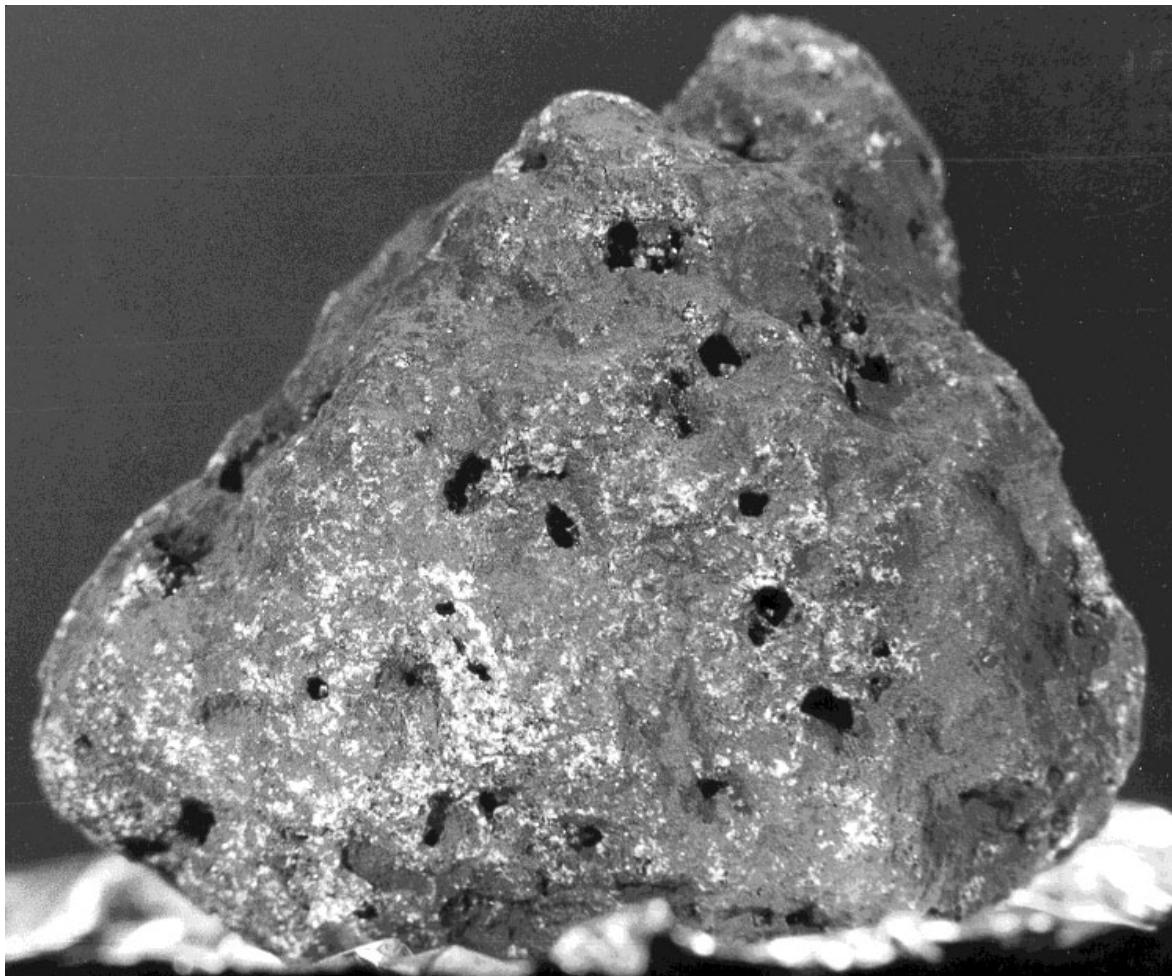


Figure 1: Photo of 10050, illustrating eroded surface. NASA S69-45731.

Introduction

Lunar Sample 10050 is a low-K, ilmenite basalt with a 3.75 b.y. crystallization age and 480 m.y. exposure age (longer than the others). Although, 10050 has been well studied, but the apparent cognate, mafic inclusions require more study.

Petrography

Schmitt et al. (1970) termed 10050 as a “coarse-grained, vuggy, ophitic cristobalite basalt.” James and Jackson (1970) grouped it with medium-grained ophitic basalts (figure 2). Frondel et al. (1970) termed 10050 a “microgabbro”. Beaty and Albee (1978) found that 10050 was unlike the other Apollo 11 basalts and deserved a category all by itself.

Beaty and Albee (1978) found that 10050,31 contains a “xenolith” whose minerals were “out of equilibrium” with the rest of the sample. The pyroxene crystals in this inclusion were more mafic $\text{Wo}_3\text{En}_{73}\text{Fs}_{21}$.

Mineralogy

Olivine: Frondel et al. (1970) and Beaty and Albee (1978) found a range in olivine composition Fo_{70-40} . Fayalite is also reported.

Pyroxene: Ross et al. (1970), Frondel et al. (1970) and Beaty and Albee (1978) reported pyroxene composition and studied chemical zoning from center

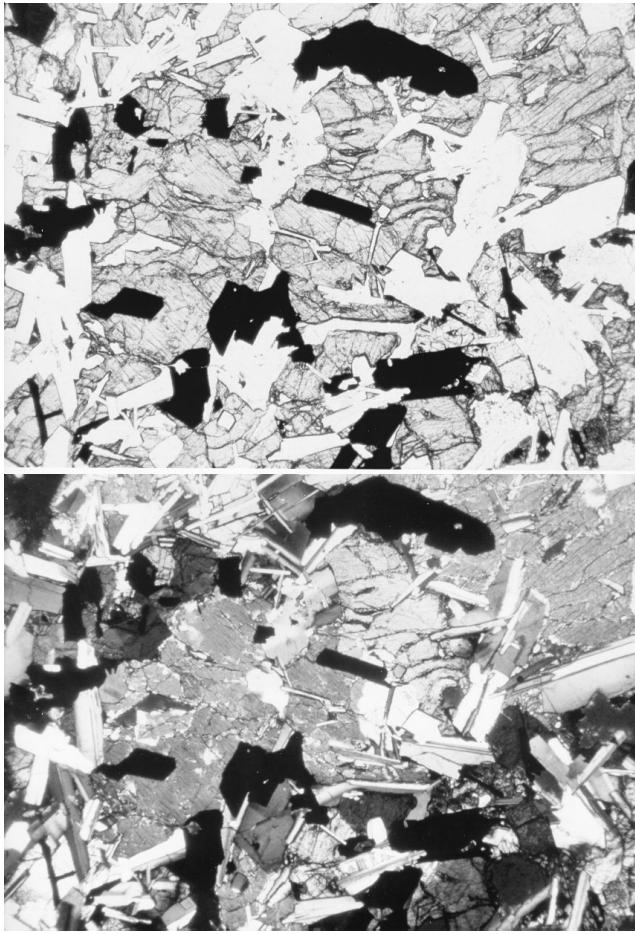


Figure 2: Photomicrograph of thin section 10050 (top: plane polarized, bottom: crossed polarized). NASA S70-49003 and 4. Scale is 2.5 mm.

to edge (figure 3). An unusual feature of the pyroxene in 10050 is that the most Fe-rich is lower in Ca than in other lunar basalts.

Plagioclase: Frondel et al. (1970) and Beatty and Albee (1978) reported the average plagioclase analysis was An_{90} and An_{92-70} , respectively (figure 4). Appleman et al. (1971) determined the cell dimensions of plagioclase from 10050 and other samples.

Ilmenite: Frondel et al. (1970) reported a “wet chemical” analysis of ilmenite from 10050.

Troilite: Evans (1970) determined the crystal structure of a euhedral troilite crystal found growing in a vug of 10050.

Chemistry

The original analyses of 10050 were by LSPET (1969), Rose et al. (1970), Goles et al. (1970), Wakita et al.

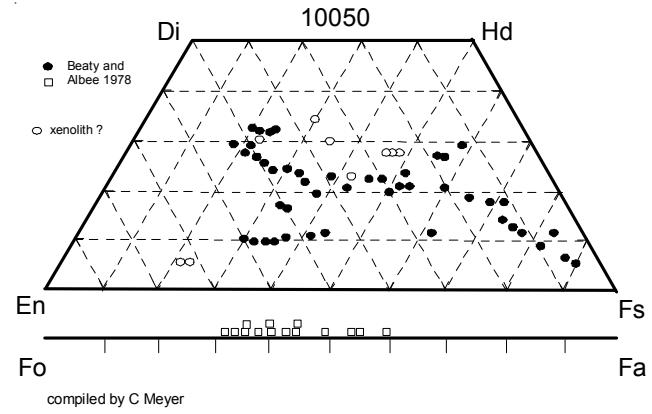


Figure 3: Composition of olivine and pyroxene in 10050.

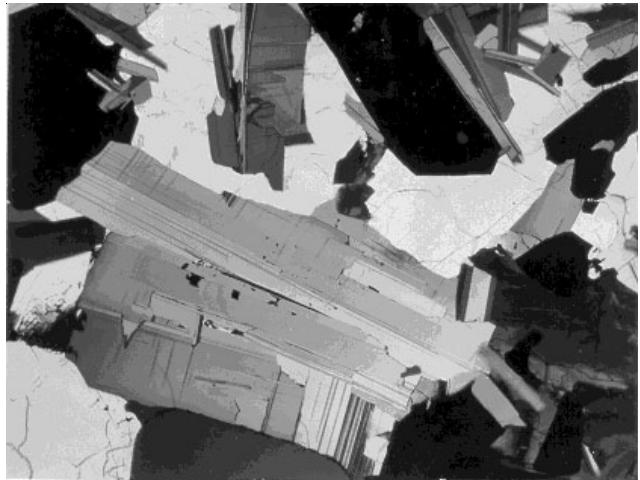


Figure 4: Large plagioclase grain in 10050. 80x

(1970), Ganapathy et al. (1970), Tera et al. (1970) and others (table 1). Rhodes and Blanchard (1980) reanalyzed the sample with similar results. It is a low K variety of Apollo 11 basalt (figures 5 and 6). Moore et al. (1970) reported an average of 64 ppm carbon and 30 ppm nitrogen.

Mineralogical Mode for 10050

	James and Jackson 70	Beatty and Albee 1978
Olivine	1.2	0.73
Pyroxene	50.2	48.8
Plagioclase	29.5	33.6
Ilmenite	14.4	14.2
Glass	0.6	0.07
silica	2.8	2.3
troilite	0.4	0.23
phosphate	0.4	0.05

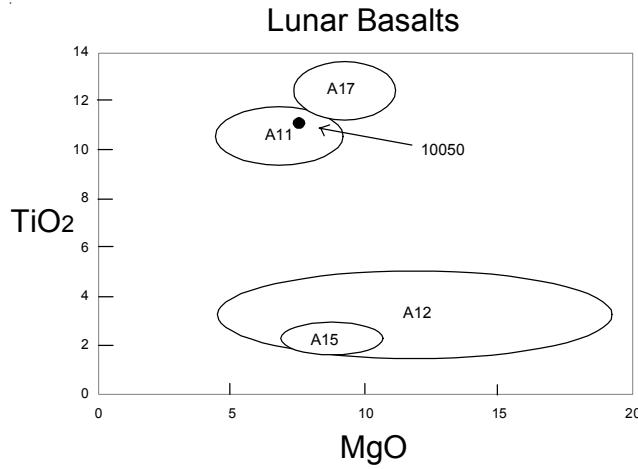


Figure 5: Composition of 10050 compared with that of other Apollo lunar samples.

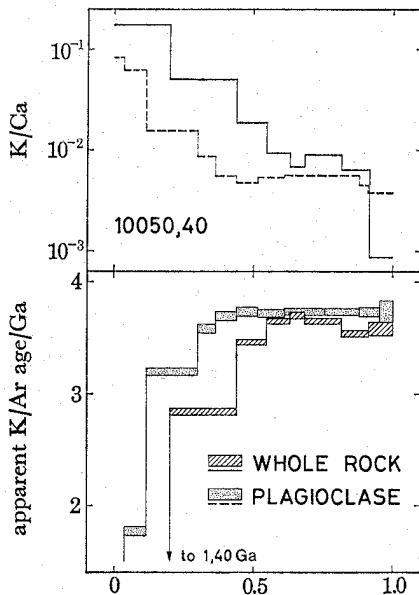


Figure 7: Argon release pattern for 10050 whole rock and plagioclase (from Geiss et al. 1977).

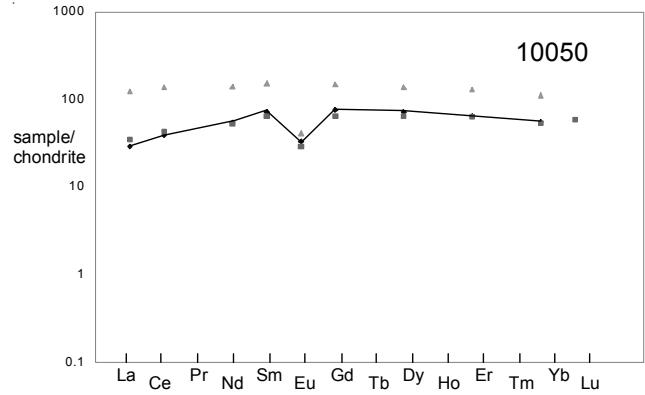


Figure 6: Normalized rare-earth-element composition for low-K basalt 10050 (the line) compared with that of low-K basalt 10020 and high-K basalt 10049 (the dots) (data from Wiesmann et al. 1975).

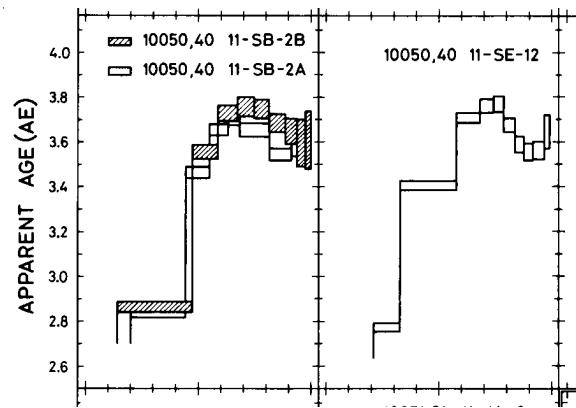


Figure 8: Argon release patterns for 10050 from Guggisberg et al. (1979).

Cosmogenic isotopes and exposure ages

Geiss et al. (1977) found that 10050 had a long exposure history compared with other low K basalts. Guggisberg et al. (1979) reported an $^{37}\text{Ar}/^{38}\text{Ar}$ cosmic ray exposure age of 480 m.y.

Other Studies

The total organic carbon content of 10050 was determined by hydrogen flame ionization pyrolysis (Ponnampерuma et al. 1970).

Funkhouser et al. (1970) and Bogard et al. (1971) reported the abundance and isotopic composition of rare gasses from 10050.

Tatsumoto et al. (1970) determined the concentration of U and Th and isotopic composition of Pb.

Summary of Age Data for 10050

	Ar/Ar (plagioclase)
Geiss et al. 1977	3.75 ± 0.03 b.y.
Guggisberg et al. 1979	3.75 ± 0.03

Table 1. Chemical composition of 10050.

reference weight	LSPET69	Wiesmann75 67 mg	Tera70	Rose70	Goles70	Rhodes80	Wakita70 600 mg	575 mg	Ganapathy70
SiO ₂ %	38	(a)		40.9	(c) 39.4	(d) 39.25	40.62	(e) 42.1	43.6 (d)
TiO ₂	9	(a)	11	(b)	12.6	(c) 11	(d) 11.79	9.61	(e) 12.8 (d)
Al ₂ O ₃	11	(a)			8.9	(c) 10.4	(d) 9.64	10.87	(e) 10.4 10.8 (d)
FeO	20	(a)			17.3	(c) 17.5	(d) 19.29	16.51	(e) 20 19 (d)
MnO	0.5	(a)			0.27	(c) 0.26	(d) 0.3	0.26	(e) 0.3 0.28 (d)
MgO	10	(a)			8	(c) 10.8	(d) 7.86	7.82	(e) 7.3 6.96 (d)
CaO	10	(a)		11.4	(b) 11.3	(c) 11.5	(d) 11.28	12.65	(e) 11.2 (d)
Na ₂ O	0.51	(a)	0.35		0.37	0.66	(c) 0.35	(d) 0.36	0.35 (d) 0.41 (d)
K ₂ O	0.06	(a)	0.065	(b)	0.08	(b) 0.05	(c)	0.08	0.06 (e) 0.05 0.06 (d)
P ₂ O ₅							0.08	0.07	(e)
S %									
<i>sum</i>									
Sc ppm					88.9	(d) 91			
V	80	(a)			117	(d)	(d) 96	89	(d)
Cr	4800	(a)		2400	(c) 2120	(d) 2415	(d) 2410	99 2230	(d)
Co	10	(a)			13.6	(d) 13.9	(d) 19	19	(d) 15.2 (f)
Ni	55	(a)							
Cu	10	(a)							15.2 (f)
Zn									1.75 (f)
Ga	8	(a)				8.8	(e)		4.41 (f)
Ge ppb									
As									
Se									
Rb	0.8	(a)	0.567	(b)	0.75	(b)	0.6	(e)	0.6 (f)
Sr	140	(a)	165	(b)	171	(b)	183	(e)	
Y	130	(a)					103	(e)	104 (f)
Zr	700	(a)	300	(b)				650	390 (d)
Nb									
Mo									
Ru									
Rh									
Pd ppb									1.4 (f)
Ag ppb									1.42 (f)
Cd ppb									2.56 (f)
In ppb									4.4 (f)
Sn ppb									
Sb ppb									
Te ppb									
Cs ppm			0.03	(b)					
Ba	60	(a)	63.5	(b)	92	(b)		70	68 (f)
La			6.77	(b)			7.2	(d) 8.9	8 (f)
Ce			23.4	(b)			34	(d) 29	(d) 37 (f)
Pr									6.2 (f)
Nd			24.8	(b)					36 (f)
Sm			10.6	(b)			11.9	(d) 12.2	12.9 (f)
Eu			1.84	(b)			2	(d) 1.92	(d) 2.2 (f)
Gd			14.9	(b)					19.9 (f)
Tb						2.1	(d) 2.5	(d)	4.3 (f)
Dy			17.6	(b)			4.6	(d)	28 (f)
Ho									4.9 (f)
Er			10.2	(b)					
Tm									
Yb	2.7	(a)	9.17	(b)			11.1	(d) 10	(d) 13.6 12.2 (d)
Lu							1.96	(d) 1.52	(d) 1.9 1.7 (d)
Hf							8.6	(d) 9.3	(d) 15 12 (d)
Ta							2.2	(d) 2.1	(d)
W ppb									
Re ppb									
Os ppb									
Ir ppb									0.24 (f)
Pt ppb									
Au ppb									0.33 (f)
Th ppm						0.7		(d) 2	1.6 (d)
U ppm			0.2	(b)					

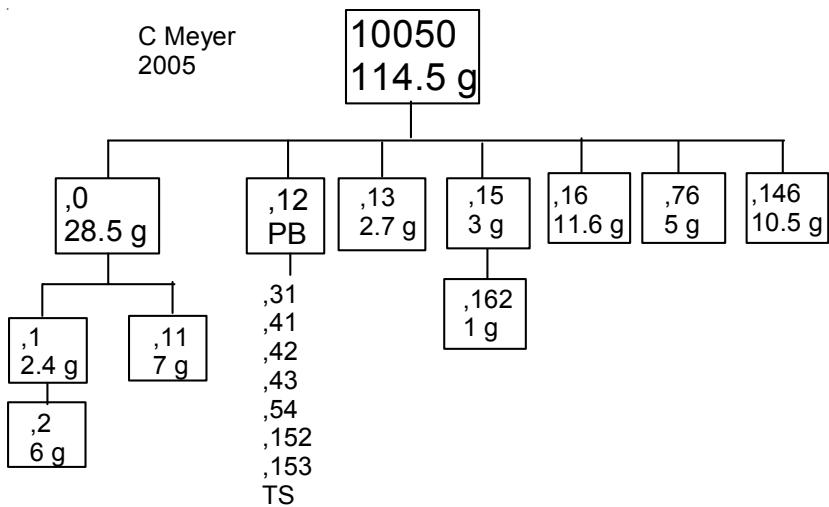
technique: (a) OES, (b) IDMS, (c) semimicro XRF, (d) INAA, (e) XRF, (f) RNAA

Table 1b. Chemical composition of 10050.

reference	Duncan 76	Beatty 78	Tatsumoto70	Gopalan70
<i>weight</i>				
SiO ₂ %	39.7	(e)	40.05	(g)
TiO ₂	11.16	(e)	11.12	(g)
Al ₂ O ₃	10.21	(e)	10.49	(g)
FeO	18.57	(e)	17.83	(g)
MnO	0.262	(e)	0.28	(g)
MgO	7.95	(e)	7.9	(g)
CaO	11.91	(e)	11.67	(g)
Na ₂ O	0.44	(e)	0.36	(g)
K ₂ O	0.052	(e)	0.04	(g)
P ₂ O ₅	0.094	(e)	0.02	(g)
S %	0.143	(e)	0.12	(g)
<i>sum</i>				
Sc ppm				
V	88	(e)		
Cr	2484	(e)	2330	(g)
Co	14	(e)		
Ni	<2	(e)		
Cu				
Zn				
Ga				
Ge ppb				
As				
Se				
Rb	<1.4	(e)		
Sr	172	(e)		
Y	103	(e)		
Zr	265	(e)		
Nb	21	(e)		
Mo				
Ru				
Rh				
Pd ppb				
Ag ppb				
Cd ppb				
In ppb				
Sn ppb				
Sb ppb				
Te ppb				
Cs ppm				
Ba	98	(e)		
La				
Ce				
Pr				
Nd				
Sm				
Eu				
Gd				
Tb				
Dy				
Ho				
Er				
Tm				
Yb				
Lu				
Hf				
Ta				
W ppb				
Re ppb				
Os ppb				
Ir ppb				
Pt ppb				
Au ppb				
Th ppm			0.531	(b)
U ppm			0.156	(b)

technique: (b) IDMS, (e) XRF, (g) elec. Probe

C Meyer
2005



Processing

Apollo 11 samples were originally described and cataloged in 1969 and “recataloged” by Kramer et al. (1977).

List of Photo #s for 10050

S69-45718 - 45	B&W mug
S69-57234	TS B&W
S70-49003 - 6	TS color
S70-50018 - 8	
S76-21738 - 9	color
S76-21349 - 51	
S76-26261	TS